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QUESTION 1 SOLUTION AND EXPLANATIONS (a)

A dative covalent or coordinate covalent bond is one in which one of the atoms supplies both the shared electrons to the covalent bond. A dative covalent bond is written $A \rightarrow B$, where the direction of the arrow shows the direction in which the

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electron pair is donated.

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U1 P2 Q1 Past Paper

Answer And

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Main Points To Be Used

In Answering A Past

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the group there is a

change in structure from

giant molecular to giant

metallic and a change in

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to metallic. From C to Ge elements exhibit a giant molecular structure. Sn and Pb exhibit a giant metallic lattice structure.

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ANSWERS. MODULE

1: FUNDAMENTALS

IN CHEMISTRY 1.

(a)(i) Define the term
bond energy [1. mark]

The bond energy

represents: The quantity

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of energy evolved when the bond is formed.

Also, the bond energy is the amount of energy absorbed when the bond is broken. (ii) The bond lengths and bond energies of carbon-carbon single, double and triple bonds are listed below.

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