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The results are: $\rho(z) = \rho_0 (1 - \beta z)$ where $\beta = \frac{\rho_0 g}{p_0} = \frac{1.20252 \text{ kg/m}^3 \cdot 9.81 \text{ m/s}^2}{101.325 \text{ kPa}} = 0.0117 \text{ 1/m}$. At $z = 7 \text{ km}$, the equation would give $\rho = 0.60 \text{ kg/m}^3$.

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Pg = 80 kPa A = 35 cm² Patm = 95 kPa mP = 4 kg 50 N Çengel: Introduction to Thermodynamics and Heat Transfer, Second Edition I. Thermodynamics 2. Introduction and Basic Concepts 56 © The McGraw-Hill Companies, 2008 pressure is 100 kPa.

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State 1 State 2 Process path Property B Property A FIGURE 2u201313 A process between states 1 and 2 and the process path. Çengel: Introduction to Thermodynamics and Heat Transfer, Second Edition I. Thermodynamics 2. Introduction and Basic Concepts 33© The McGraw-Hill Companies, 2008 process completely, one should specify the initial and final states of the process, as well as the path it follows, and the interactions with the surroundings.

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The efficiency of a cook- ing appliance is defined as the ratio of the useful energy transferred to the overall h combustionthermalgenerator u00fd W # net,electric HHV m# net 86 | Introduction to Thermodynamics and Heat Transfer TABLE 3u20131 The efficacy of different lighting systems Efficacy, Type of lighting lumens/W Combustion Candle 0.2 Incandescent Ordinary 6u201320 Halogen ...

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