

Chapter 13 Chemistry Answers

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Answers to Chapter 13 Study Questions. 1. True: b), c) and d) 2. Reaction = 2 x reverse of original reaction. So $K_2 = (1/K_1)^2 = (1/0.447)^2 = 5.00$. 3. $K = [NH_3]^2 / [N_2] \times [H_2]^3$; more product made. Increasing the pressure increases the concentration; counteract this by shifting to the side with fewer moles.

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Answer Key Chapter 13 - Chemistry 2e | OpenStax. 1. The reaction can proceed in both the forward and reverse directions. 3. When a system has reached equilibrium, no further changes in the reactant and product concentrations occur; the forward and reverse reactions continue to proceed, but at equal rates. 5.

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Answers to Chemistry End of Chapter Exercises. 1. The reaction can proceed in both the forward and reverse directions. 3. When a system has reached equilibrium, no further changes in the reactant and product concentrations occur; the reactions continue to occur, but at equivalent rates. 5.

13.1 Chemical Equilibria – Chemistry
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Answer/Explanation. Answer: d. Explanation: 4. An organic compound C₇H₉N forms clear solution when dissolved in KOH after reacting with C₆H₅SO₂Cl, 'A' on diazotisation at 0°C and then reaction with ?-naphthol gives orangish red dye. 'A' on electrophilic substitution gives single product. 'A' is.

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Chemistry EDEXCEL INTERNATIONAL GCSE CHEMISTRY EDEXCEL CERTIFICATE IN CHEMISTRY ANSWERS SECTION A Chapter 1 1. a) melting b) freezing c) subliming / sublimation d) subliming / sublimation 2. a) Note: Solids should have regularly packed particles touching. Liquids should have most of the

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